

## Ingredients and cytotoxicity of MTA and 3 kinds of Portland cements

Seok-Woo Chang<sup>1</sup>, Hyun-Mi Yoo<sup>1</sup>, Dong Sung Park<sup>1</sup>, Tae-Seok Oh<sup>1</sup>, Kwang-Shik Bae<sup>2,3\*</sup>

<sup>1</sup>*Department of Conservative Dentistry, The Institute of Oral Health Science, Samsung Medical Center,*

*Sungkyunkwan University School of Medicine*

<sup>2</sup>*Department of Conservative Dentistry, School of Dentistry, Seoul National University*

<sup>3</sup>*Dental Research Institute, Seoul National University*

### ABSTRACT

The aim of this study was to compare the compositions and cytotoxicity of white ProRoot MTA (white mineral trioxide aggregate) and 3 kinds of Portland cements. The elements, simple oxides and phase compositions of white MTA (WMTA), gray Portland cement (GPC), white Portland cement (WPC) and fast setting cement (FSC) were measured by inductively coupled plasma atomic emission spectrometry (ICP-AES), X-ray fluorescence spectrometry (XRF) and X-ray diffractometry (XRD). Agar diffusion test was carried out to evaluate the cytotoxicity of WMTA and 3 kinds of Portland cements.

The results showed that WMTA and WPC contained far less magnesium (Mg), iron (Fe), manganese (Mn), and zinc (Zn) than GPC and FSC. FSC contained far more aluminum oxide ( $\text{Al}_2\text{O}_3$ ) than WMTA, GPC, and WPC. WMTA, GPC, WPC and FSC were composed of main phases, such as tricalcium silicate ( $3\text{CaO} \cdot \text{SiO}_2$ ), dicalcium silicate ( $2\text{CaO} \cdot \text{SiO}_2$ ), tricalcium aluminate ( $3\text{CaO} \cdot \text{Al}_2\text{O}_3$ ), and tetracalcium aluminoferrite ( $4\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Fe}_2\text{O}_3$ ). The significance of the differences in cellular response between WMTA, GPC, WPC and FSC was statistically analyzed by Kruskal-Wallis Exact test with Bonferroni's correction. The result showed no statistically significant difference ( $p > 0.05$ ).

WMTA, GPC, WPC and FSC showed similar compositions. However there were notable differences in the content of minor elements, such as aluminum (Al), magnesium, iron, manganese, and zinc. These differences might influence the physical properties of cements. [J Kor Acad Cons Dent 33(4):369-376, 2008]

**Key words :** White MTA (white mineral trioxide aggregate), Portland cement, ICP-AES (Inductively coupled plasma atomic emission spectrometry), XRF (X-ray fluorescence spectrometry), XRD (X-ray diffractometry), Agar diffusion test

- Received 2008.2.16., revised 2008.2.28., accepted 2008.6.5.-

---

\* Corresponding Author: **Kwang-Shik Bae**

Department of Conservative Dentistry,  
School of Dentistry, Seoul National University  
28 Yeongeon-dong, Jongro-gu, Seoul, 110-749, Korea  
Tel: +82-2-2072-2650 Fax: +82-2-2072-3859  
E-mail: address: baeks@snu.ac.kr

### I . INTRODUCTION

Mineral trioxide aggregate (MTA) was developed in 1993 and has expanded its application widely. Because of its good biocompatibility<sup>1,2)</sup>, physical property, and antibacterial effect<sup>3)</sup>, it has been

※This work was supported by grant no. 04-2007-0020 from the SNUHD Research Fund.

used in root end filling<sup>4,6)</sup>, pulpotomy<sup>7)</sup>, perforation repair<sup>8,9)</sup>, coronal barrier<sup>10)</sup>, and root canal filling.

In spite of these superior characteristics, the high cost of MTA limited its use in Korea. Recently many studies have been done to compare the compositions<sup>11-13)</sup>, biocompatibility<sup>14-16)</sup>, and physical properties<sup>17-19)</sup> of MTA and Portland cements. Many researchers suggested that Portland cement is as biocompatible and have as good physical properties as MTA.

But Dammaschke *et al.*<sup>20)</sup> reported that MTA contained significantly less iron (Fe) and manganese (Mn) than Portland cement and MTA cannot be simply replaced by cheaper Portland cement. But so far, only a little information about chemical compositions of Portland cement can be found in the literature<sup>20)</sup> and precise comparative study of the compositions of MTA and Portland cement is required.

The main drawback of MTA is its long setting time. The setting time of MTA was reported to be 2 h 45 min<sup>21)</sup>. Experimentally, CaCl<sub>2</sub> was added to shorten the setting time and improve the mechanical properties of MTA<sup>22,23)</sup>. In this experiment, fast setting cement was used as future possible alternative for MTA. White Portland cement was also used in this experiment, which is known to be the material used for the production of white MTA<sup>24)</sup>.

The purposes of this study were to investigate the compositions and evaluate the cytotoxicity of MTA (tooth colored formula, WMTA), gray Portland cement (GPC), white Portland cement (WPC), and fast setting cement (FSC).

## II. MATERIALS AND METHODS

We used GPC (Lafarge Halla cement Corp., Seoul, Korea), WPC (Union Corp., Seoul, Korea), FSC (SSangyong cement industrial Co., Seoul, Korea) and WMTA (ProRoot MTA, tooth colored formula, Dentsply Tulsa dental, Johnson city, TN, USA).

### Analysis of element by ICP-AES (Inductively coupled plasma atomic emission spectrometry)

ICP-AES (ICPS-1000Ⅳ, Shimadzu, Kyoto, Japan) was used to investigate the elements that comprise WMTA, GPC, WPC and FSC. Argon Plasma (6000K) was used. Detection limit of ICP-AES was 1-10 ppb. Measurement was carried out in triplicate and the mean value was determined.

### Analysis of simple oxide by XRF (X-ray fluorescence spectrometry)

XRF (XRF-1700, Shimadzu, Kyoto, Japan) was used to investigate the relative proportions of simple oxides that comprise WMTA, GPC, WPC and FSC. We used Rh (rhodium) target and selected 40Kv and 30 mA for X-ray generator. Measurement was carried out in triplicate and the mean value was determined.

### Phase identification by XRD (X-ray diffraction)

To investigate the large compounds and phases, XRD analysis (D8-advance, Bruker, Madison, WI, USA) was carried out. Phase identification was accomplished by the use of search-match software (Eva Version 9.0 Diffrac plus, Bruker, Madison, WI, USA). Target material was Cu. System capacity was 40kV and 40 mA.

XRF analysis was carried out in NCIRF (National Center for Inter-University Research Facilities, Seoul, Korea). XRD analysis was carried out in SNU DRI (Seoul National University Dental Research Institute, Seoul, Korea). ICP-AES analysis was carried out in NICEM (National Instrumentation center for environmental management, Seoul, Korea).

### Cytotoxicity test by Agar diffusion method

L929 mouse fibroblast cells were used. Cells were grown in minimum essential medium under standard cell culture conditions (37°C, 5 % CO<sub>2</sub>). Experimental materials were sterilized with ethylene oxide gas. Cultured cells were seeded into 6-well plate at an initial density of  $2 \times 10^5$  cells/well with 2 ml of medium and incubated for 24 hours. After 24 hours, 2 ml mixture of agar-

medium was added into each well of 6-well plate and stained with Neutral Red solution. Freshly mixed MTA (water powder [w/p] ratio 1 : 1), GPC (w/p ratio 1 : 2), WPC (w/p ratio 1 : 2) and FSC (w/p ratio 1 : 2) were inserted in polyethylene tube (3 mm in inner diameter and 5 mm in height) and placed on cultured cells. For each material (WMTA, GPC, WPC, and FSC), agar diffusion test was carried out in triplicate and the mean value was determined.

6, 12, 18, and 24 hours after placing the samples, the width of decolorization zones was measured. Zinc oxide eugenol (ZOE) paste and Teflon discs were used as positive and negative controls. The significance of the difference in cellular response between WMTA, GPC, WPC and FSC was statistically analyzed by Kruskal-Wallis Exact test with Bonferroni's correction.

### III. RESULTS

#### Compositions of elements

The compositions of main elements were shown in Table 1. FSC showed remarkably high concentrations of aluminum (Al), sulfur (S), and titanium (Ti). WPC showed high concentration of Fluorine (F). The concentrations of iron, magnesium (Mg), manganese (Mn) and zinc (Zn) in

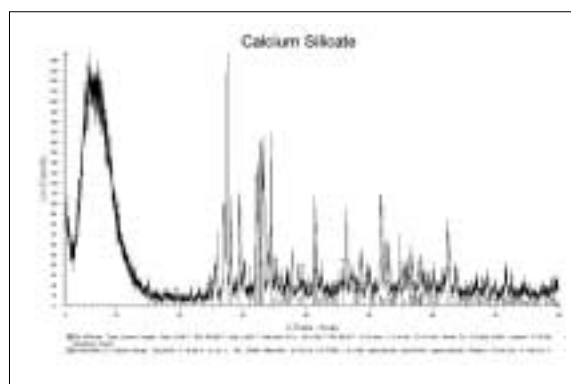
WMTA and WPC were lower than those in GPC and FSC.

#### Compositions of main oxides

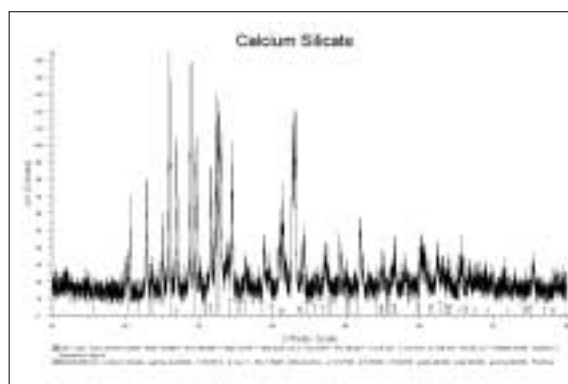
The results of XRF showed that calcium oxide (CaO) and silicate ( $\text{SiO}_2$ ) were two main simple oxides that comprise WMTA, GPC, WPC and FSC. Aluminum oxide ( $\text{Al}_2\text{O}_3$ ) content in FSC was higher than those in WMTA, GPC and WPC. The compositions of main oxides that comprise WMTA and Portland cements were shown in Table 1.

#### Mineral phase identification

The results of XRD analysis showed that main mineral phases comprising WMTA and Portland cements were similar (Figure 1). They were mainly tricalcium silicate ( $3\text{CaO} \cdot \text{SiO}_2$ ), dicalcium silicate ( $2\text{CaO} \cdot \text{SiO}_2$ ), tricalcium aluminate ( $3\text{CaO} \cdot \text{Al}_2\text{O}_3$ ) and tetracalcium aluminoferrite ( $4\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Fe}_2\text{O}_3$ ). Other mineral phases such as calcium silicate ( $3\text{CaO} \cdot \text{SiO}_2$ ), calcium aluminum oxide ( $3\text{CaO} \cdot \text{Al}_2\text{O}_3$ ), calcium aluminum oxide sulfate ( $3\text{CaO} \cdot 3\text{Al}_2\text{O}_3 \cdot \text{CaSO}_4$ ), magnesium oxide (MgO), cummingtonite ( $(\text{Mg}_{4.68}\text{Fe}_{2.32})\text{Si}_8\text{O}_{22}(\text{OH})_2$ ), magnesiocummingtonite ( $(\text{Fe}_{3.17}\text{Mg}_{3.83})(\text{Si}_8\text{O}_{22}(\text{OH})_2)$ ), gehlenite ( $2\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{SiO}_2$ ) were also found in WMTA and Portland cements.



(a) White MTA



(b) Portland cement

**Figure 1.** X-ray diffraction (XRD) analysis of white MTA (a) and Portland cement (b) showing the main phase ( $2\text{CaO} \cdot \text{SiO}_2$ ) present in the cement. (MTA: mineral trioxide aggregate) (Black : peaks made by specimen, Red : peaks of reference)

**Table 1.** Compositions of elements, simple oxides, and phases that comprise WMTA, GPC, WPC and FSC

		MTA	GPC	WPC	FSC
Composition of main elements measured by ICP -AES(wt %)	Ca	25.23	29.33	29.36	23.47
	S	0.87	1.17	1.41	4.30
	Al	0.71	2.33	2.22	8.47
	Mg	0.25	1.88	0.65	1.01
	Fe	0.11	1.42	0.13	1.03
	K	0.09	0.72	0.10	0.42
	Sr	0.05	0.03	0.04	0.02
	F	0.0225	0.0379	0.1704	0.0954
	Ti	0.01	0.06	0.02	0.23
	Mn	0.006	0.043	0.005	0.035
	Zn	0.0006	0.069	0.0009	0.0157
Composition of simple oxide measured by XRF (wt %)	CaO	47.18	61.84	65.92	48.64
	SiO <sub>2</sub>	19.42	21.00	21.39	11.62
	Al <sub>2</sub> O <sub>3</sub>	1.39	5.07	4.39	17.45
	Fe <sub>2</sub> O <sub>3</sub>	0.70	2.76	0.80	2.09
	MgO	0.34	3.28	1.16	1.50
	K <sub>2</sub> O	0.04	1.11	-	0.61
Composition of phases (wt %) calculated by Bogue's method	3CaO · SiO <sub>2</sub>		54.07	75.09	
	2CaO · SiO <sub>2</sub>		19.50	4.77	
	3CaO · Al <sub>2</sub> O <sub>3</sub>		8.77	10.28	
	4CaO · Al <sub>2</sub> O <sub>3</sub> · Fe <sub>2</sub> O <sub>3</sub>		8.39	2.43	

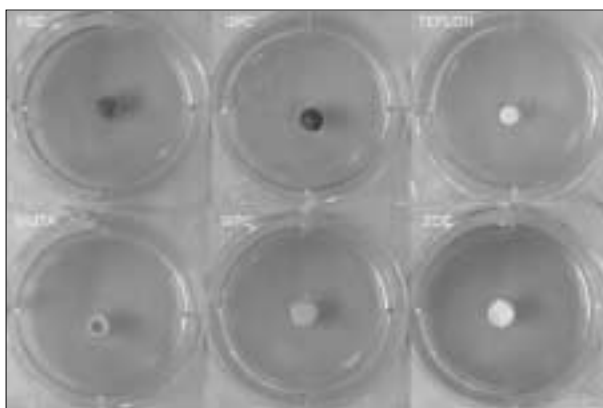
Silicon (Si) was not measured because of technical difficulty. Main phases that comprise MTA and FSC were identified by XRD analysis. But the phase compositions of MTA and FSC could not be calculated.

(WMTA: white mineral trioxide aggregate, GPC: gray Portland cement, WPC: white Portland cement, FSC: fast setting cement, XRD: X-ray diffractometry, ICP-AES: Inductively coupled plasma atomic emission spectrometry, XRF: X-ray fluorescence spectrometry)

**Table 2.** The width of decolorization zone created at 6, 12, 18, and 24 hours after placement of samples (mm)

	6H	12H	18H	24H
WMTA (n = 3)	1.23 (± 0.12)	1.37 (± 0.06)	1.43 (± 0.06)	1.5 (± 0.10)
GPC (n = 3)	1.23 (± 0.12)	1.47 (± 0.06)	1.47 (± 0.06)	1.47 (± 0.06)
WPC (n = 3)	1.23 (± 0.12)	1.43 (± 0.06)	1.47 (± 0.06)	1.47 (± 0.06)
FSC (n = 3)	1.23 (± 0.12)	1.3 (± 0.10)	1.33 (± 0.12)	1.43 (0.06)
ZOE (n = 3)	3.9	+	+	+
Teflon (n = 3)	-	-	-	-

(Mean ± SD, WMTA : White mineral trioxide aggregate, GPC : gray Portland cement, WPC : white Portland cement, FSC : fast setting cement, H : hours, + : complete decolorization, - : no discoloration)



**Figure 2.** Cellular response to WMTA, GPC, WPC, and FSC, 6 hrs after placement of specimens.

(WMTA : white MTA, GPC : Gray Portland cement, WPC : white Portland cement, and FSC : Fast setting cement, ZOE : Zinc Oxide and Eugenol)

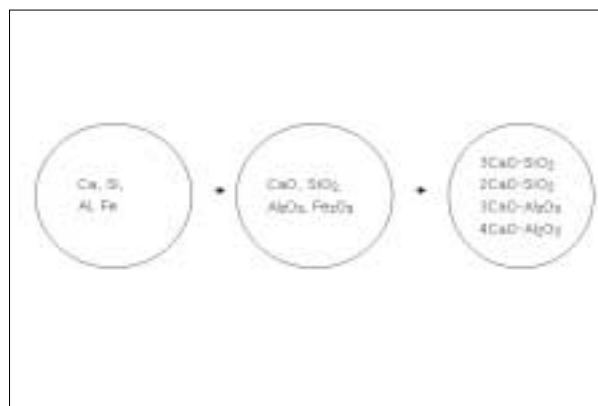
### Measurement of decolorization zone

Decolorization zones were observed around the samples. Decolorization zones were smaller than 2 mm in all groups except positive control (ZOE) group (Table 2) (Figure 2). The result showed no statistically significant difference ( $p > 0.05$ ).

## IV. DISCUSSION

Dammaschke *et al.*<sup>20)</sup> analyzed the compositions of MTA and Portland cements by X-ray photoelectron spectroscopy (XPS), energy dispersive X-ray analysis (EDX), and inductively coupled plasma optical emission spectrometry (ICP-OES). Camilleri *et al.*<sup>25,26)</sup> studied the chemical composition with EDX and XRD. These studies were mainly focused in elemental analysis and phase identification.

MTA and Portland cement are complex compounds and they are composed of mineral phases. These mineral phases are composed of simple oxides and these simple oxides are composed of elements. So in order to investigate the chemical compositions of WMTA and Portland cements, compositions of elements that comprise simple oxides were investigated. And then, compositions of simple oxides that comprise mineral phases



**Figure 3.** Schematic diagram showing the compositions of WMTA and Portland cements.

(WMTA : white mineral trioxide aggregate)

were measured (Figure 3). And finally, compositions of large phases that comprise WMTA and Portland cements were calculated.

There has been no study that investigated the compositions of WMTA and Portland cements from elemental level to mineral phase level. Knowing the compositions of WMTA and Portland cement from elemental level to mineral phase level can be helpful in understanding the physical properties of WMTA and Portland cement.

ICP-AES analysis showed that the concentrations of magnesium, iron, manganese and zinc in WMTA and WPC were remarkably lower than those in GPC and FSC. This explains the color differences between WPC and GPC because iron and manganese are well-known chromophores<sup>20)</sup>. It was interesting that WPC had the highest fluorine content among 4 samples. We might expect some anticariogenic effect of WPC when it is used clinically. Bismuth was found only in WMTA. Bismuth was reported to decrease the mechanical strength of MTA<sup>27)</sup>.

There has been few study that measured the compositions of simple oxides that comprise WMTA and Portland cements in unhydrated forms. By knowing the compositions of simple oxides, we can calculate the compositions of phas-

es that comprise Portland cement. Two main oxides that compose WMTA and Portland cements were calcium oxide and silicate. Aluminum oxide content was remarkably high in fast setting cement, which might explain the short setting time of FSC. Bismuth oxide was known to be present in WMTA. However the amount of Bismuth oxide was not measured in this study due to technical difficulties.

The compositions of phases can be calculated from the compositions of CaO, SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, Fe<sub>2</sub>O<sub>3</sub>, and SO<sub>3</sub> measured by XRF analysis. By Bogue's method<sup>28)</sup>, the compositions of tricalcium oxide, dicalcium oxide, tricalcium aluminate and tetra-calcium aluminoferrite which comprise GPC and WPC were calculated. Bogue's method was not applicable to the calculation of mineral phase of WMTA and FSC, because Bogue's method is based on the chemical assumption that Portland cement is composed entirely of CaO, SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, and Fe<sub>2</sub>O<sub>3</sub> and this assumption does not apply to WMTA and FSC.

Min *et al.*<sup>29)</sup> reported that Portland cement was biocompatible and had the potential to be used in pulp capping. Cellular responses to WMTA, GPC, WPC, and FSC showed no statistically significant differences ( $p > 0.05$ ).

## V. CONCLUSION

The main elemental composition of WMTA and 3 Portland cement were similar. WMTA and WPC contained far less magnesium, iron, manganese, and zinc than GPC and FSC. WPC contained higher Fluorine than WMTA, GPC and FSC. FSC contained higher aluminum oxide than WMTA, GPC, and WPC. WMTA, GPC, WPC, and FSC were composed of tricalcium silicate, dicalcium silicate, tricalcium aluminate, and tetracalcium aluminoferrite. The difference in cellular responses between WMTA, GPC, WPC and FSC was not statistically significant ( $p > 0.05$ ). WPC showed remarkably lower heavy metal contents and higher fluorine content than GPC and FSC. This may contribute to good biocompatibility and anticariogenic effect.

## REFERENCES

1. Torabinejad M, Ford TR, Abedi HR, Kariyawasam SP, Tang HM. Tissue reaction to implanted root-end filling materials in the tibia and mandible of guinea pigs. *J Endod* 24:468-71, 1998.
2. Koh ET, Torabinejad M, Pitt Ford TR, Brady K, McDonald F. Mineral trioxide aggregate stimulates a biological response in human osteoblasts. *J Biomed Mater Res* 37:432-9, 1997.
3. Al-Hezaimi K, Al-Shalan TA, Naghshbandi J, Oglesby S, Simon JH, Rotstein I. Antibacterial effect of two mineral trioxide aggregate (MTA) preparations against *Enterococcus faecalis* and *Streptococcus sanguis* in vitro. *J Endod* 32:1053-6, 2006.
4. Torabinejad M, Watson TF, Pitt Ford TR. Sealing ability of a mineral trioxide aggregate when used as a root end filling material. *J Endod* 19:591-5, 1993.
5. Torabinejad M, Rastegar AF, Kettering JD, Pitt Ford TR. Bacterial leakage of mineral trioxide aggregate as a root-end filling material. *J Endod* 21:109-12, 1995.
6. Torabinejad M, Smith PW, Kettering JD, Pitt Ford TR. Comparative investigation of marginal adaptation of mineral trioxide aggregate and other commonly used root-end filling materials. *J Endod* 21:295-9, 1995.
7. Torabinejad M, Chivian N. Clinical applications of mineral trioxide aggregate. *J Endod* 25:197-205, 1999.
8. Lee SJ, Monsef M, Torabinejad M. Sealing ability of a mineral trioxide aggregate for repair of lateral root perforations. *J Endod* 19:541-4, 1993.
9. Pitt Ford TR, Torabinejad M, McKendry DJ, Hong CU, Kariyawasam SP. Use of mineral trioxide aggregate for repair of furcal perforations. *Oral Surg Oral Med Oral Pathol Oral Radiol Endod* 79:756-63, 1995.
10. Tselnik M, Baumgartner JC, Marshall JG. Bacterial leakage with mineral trioxide aggregate or a resin-modified glass ionomer used as a coronal barrier. *J Endod* 30:782-4, 2004.
11. Song JS, Mante FK, Romanow WJ, Kim S. Chemical analysis of powder and set forms of Portland cement, gray ProRoot MTA, white ProRoot MTA, and gray MTA-Angelus. *Oral Surg Oral Med Oral Pathol Oral Radiol Endod* 102:809-15, 2006.
12. Islam I, Chng HK, Yap AU. X-ray diffraction analysis of mineral trioxide aggregate and Portland cement. *Int Endod J* 39:220-5, 2006.
13. Camilleri J, Montesin FE, Di Silvio L, Pitt Ford TR. The chemical constitution and biocompatibility of accelerated Portland cement for endodontic use. *Int Endod J* 38:834-42, 2005.
14. De Deus G, Ximenes R, Gurgel-Filho ED, Plotkowski MC, Coutinho-Filho T. Cytotoxicity of MTA and Portland cement on human ECV 304 endothelial cells. *Int Endod J* 38:604-9, 2005.
15. Ribeiro DA, Duarte MA, Matsumoto MA, Marques ME, Salvadori DM. Biocompatibility in vitro tests of mineral trioxide aggregate and regular and white Portland cements. *J Endod* 31:605-7, 2005.
16. de Morais CA, Bernardini N, Garcia RB, Duarte MA, Guerisoli DM. Evaluation of tissue response to MTA and Portland cement with iodoform. *Oral Surg Oral Med Oral Pathol Oral Radiol Endod* 102:417-21, 2006.
17. Santos AD, Moraes JC, Araujo EB, Yukimitu K, Valerio Filho WV. Physico-chemical properties of MTA

- and a novel experimental cement. *Int Endod J* 38:443-7, 2005.
18. Islam I, Chng HK, Yap AU. Comparison of the physical and mechanical properties of MTA and portland cement. *J Endod* 32:193-7, 2006.
19. De-Deus G, Petruccelli V, Gurgel-Filho E, Coutinho-Filho T. MTA versus Portland cement as repair material for furcal perforations: a laboratory study using a polymicrobial leakage model. *Int Endod J* 39:293-8, 2006.
20. Dammaschke T, Gerth HU, Zuchner H, Schafer E. Chemical and physical surface and bulk material characterization of white ProRoot MTA and two Portland cements. *Dent Mater* 21:731-8, 2005.
21. Torabinejad M, Hong CU, McDonald F, Pitt Ford TR. Physical and chemical properties of a new root-end filling material. *J Endod* 21:349-53, 1995.
22. Antunes Bortoluzzi E, Juarez Broon N, Antonio Hungaro Duarte M, de Oliveira Demarchi AC, Monteiro Bramante C. The use of a setting accelerator and its effect on pH and calcium ion release of mineral trioxide aggregate and white Portland cement. *J Endod* 32:1194-7, 2006.
23. Bortoluzzi EA, Broon NJ, Bramante CM, Garcia RB, de Moraes IG, Bernardineli N. Sealing ability of MTA and radiopaque Portland cement with or without calcium chloride for root-end filling. *J Endod* 32:897-900, 2006.
24. Tay FR, Pashley DH, Rueggeberg FA, Loushine RJ, Weller RN. Calcium Phosphate Phase Transformation Produced by the Interaction of the Portland Cement Component of White Mineral Trioxide Aggregate with a Phosphate-containing Fluid. *J Endod* 33:1347-51, 2007.
25. Camilleri J, Montesin FE, Brady K, Sweeney R, Curtis RV, Ford TR. The constitution of mineral trioxide aggregate. *Dent Mater* 21:297-303, 2005.
26. Camilleri J, Montesin FE, Curtis RV, Ford TR. Characterization of Portland cement for use as a dental restorative material. *Dent Mater* 22:569-75, 2006.
27. Coomaraswamy KS, Lumley PJ, Hofmann MP. Effect of bismuth oxide radioopacifier content on the material properties of an endodontic Portland cement-based (MTA-like) system. *J Endod* 33:295-8, 2007.
28. Camilleri J. Hydration mechanisms of mineral trioxide aggregate. *Int Endod J* 40:462-70, 2007.
29. Min KS, Kim HI, Park HJ, Pi SH, Hong CU, Kim EC. Human pulp cells response to Portland cement in vitro. *J Endod* 33:163-6, 2007.

## 국문초록

### MTA와 포틀랜드 시멘트의 구성성분분석과 세포독성에 관한 연구

장석우<sup>1</sup> · 유현미<sup>1</sup> · 박동성<sup>1</sup> · 오태석<sup>1</sup> · 배광식<sup>2,3\*</sup>

<sup>1</sup>성균관대학교 의과대학 삼성서울병원 치과보존과

<sup>2</sup>서울대학교 치의학대학원 치과보존학교실, <sup>3</sup>서울대학교 치학연구소

이 연구의 목적은 3종의 포틀랜드 시멘트 (포틀랜드 시멘트, 백색 포틀랜드 시멘트, 초속경 시멘트)와 white MTA의 성분 및 세포독성을 비교하는 것이다. 성분비교를 위해서 X선 회절기 (XRD), X선 형광분석기 (XRF), 유도결합플라즈마 원자방출분광 분석기 (ICP-AES)를 사용하였으며, 세포독성비교를 위해서는 우무확산법 (agar diffusion test)을 사용하였다. 분석 결과, white MTA와 백색 포틀랜드 시멘트는 포틀랜드 시멘트나 초속경 시멘트에 비해 적은 양의 마그네슘 (mg), 철 (Fe), 아연 (Zn), 그리고 망간 (Mn)을 함유하고 있었다. 또한 초속경 시멘트는 다른 시멘트 및 white MTA에 비해 많은 산화 알루미늄 ( $\text{Al}_2\text{O}_3$ )을 함유하고 있었다. MTA와 포틀랜드 시멘트의 주된 성분은 tricalcium silicate ( $3\text{CaO} \cdot \text{SiO}_2$ ), dicalcium silicate ( $2\text{CaO} \cdot \text{SiO}_2$ ), tricalcium aluminate ( $3\text{CaO} \cdot \text{Al}_2\text{O}_3$ ), 그리고 tetracalcium aluminoferrite ( $4\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Fe}_2\text{O}_3$ )들이었다. 세포독성 실험결과를 Kruskal-Wallis Exact test와 Bonferroni 사후 검정법을 사용하여 분석한 결과 white MTA와 3종의 포틀랜드 시멘트 군 사이에서 통계적으로 유의성 있는 차이를 보이지 않았다 ( $p > 0.05$ ). White MTA와 3종의 포틀랜드 시멘트의 주성분은 유사하였으나 알루미늄 (Al), 마그네슘 (mg), 철 (Fe), 아연 (Zn), 그리고 망간 (Mn) 등의 함량에서는 차이를 보였으며 이러한 차이들은 물리적 성질에 영향을 미칠 것으로 보인다.

**주요어:** MTA (mineral trioxide aggregate), 포틀랜드 시멘트, XRD, XRF, ICP-AES, 우무확산법